Bioorganic & Medicinal Chemistry Letters 16 (2006) 4922-4930

Bioorganic & Medicinal Chemistry Letters

Synthesis and structure—activity relationships of retro bis-aminopyrrolidine urea (rAPU) derived small-molecule antagonists of the melanin-concentrating hormone receptor-1 (MCH-R1). Part 2

Sarah Hudson,^{a,*} Mehrak Kiankarimi,^a Martin W. Rowbottom,^a Troy D. Vickers,^a Dongpei Wu,^a Joseph Pontillo,^a Brett Ching,^a Wesley Dwight,^a Val S. Goodfellow,^a David Schwarz,^b Christopher E. Heise,^c Ajay Madan,^d Jenny Wen,^d William Ban,^d Hua Wang^d and Warren S. Wade^{a,*}

^aDepartment of Medicinal Chemistry, Neurocrine Biosciences Inc., 12790 El Camino Real, San Diego, CA 92130, USA
 ^bDepartment of Molecular Biology, Neurocrine Biosciences Inc., 12790 El Camino Real, San Diego, CA 92130, USA
 ^cDepartment of Pharmacology, Neurocrine Biosciences Inc., 12790 El Camino Real, San Diego, CA 92130, USA
 ^dDepartment of Preclinical Development, Neurocrine Biosciences Inc., 12790 El Camino Real, San Diego, CA 92130, USA

Received 16 May 2006; revised 13 June 2006; accepted 14 June 2006 Available online 7 July 2006

Abstract—The design, synthesis, and SAR of a series of retro bis-aminopyrrolidine ureas are described. Compounds from this series exhibited considerable binding affinity ($K_i = 1 \text{ nM}$) and functional activity at MCH-R1, acceptable CYP2D6 inhibition, and good rat brain exposure.

© 2006 Elsevier Ltd. All rights reserved.

Mammalian melanin-concentrating hormone (MCH) is a 19 amino acid cyclic peptide¹ which selectively binds and activates two 7-transmembrane G protein-coupled receptors, namely MCH-R1 and R2.² Recent research suggests that MCH-R1 is involved in the modulation of energy homeostasis, food intake, and body weight.³ Animal models and other data suggest that, in man, blockade of MCH-R1 may lead to a clinical treatment for chronic obesity. A number of groups have recently reported on their efforts toward the development of selective small-molecule MCH-R1 antagonists.^{3,4}

In the previous paper we discussed our efforts to optimize the retro bis-aminopyrrolidine urea scaffold (rAPU), a series of potent and functional MCH-R1 antagonists.⁵ This series was derived from the bisaminopyrrolidine ureas (APU)⁴ⁱ and had significantly

Keywords: Retro bis-aminopyrrolidine urea; Melanin-concentrating hormone receptor-1; Antagonists; CYP2D6.

less activity in the hERG patch-clamp assay. The series also exhibited greater metabolic stability in human liver microsomes which translated to greater oral bioavailability in rat. Experiments to investigate the SAR of the terminal aryl ring of the biaryl moiety revealed that para lipophilic substitution was favored, resulting in a dramatic increase in MCH-R1 potency. However, two problems were identified: significant CYP2D6 inhibition and low brain penetration. With these issues in mind, the left- and right-hand sides of the rAPU were modified to identify compounds equipotent to APU 1 (MCH-R1 $K_i = 2$ nM) with the same low liability for inhibition of CYP2D6, while retaining the superior properties demonstrated by previous rAPUs (Fig. 1).

Figure 1. Aminopyrrolidine urea 1.

^{*}Corresponding authors. Tel.: +1 858 617 7600; fax: +1 858 617 7601; e-mail: shudson@neurocrine.com

To increase the efficiency of functionalizing both ends of the rAPU core, the previously reported synthesis⁵ was modified to generate the key intermediate 6 (Scheme 1). (3S)-(-)-1-Benzyl-3-(methylamino)pyrrolidine $(2)^6$ was first protected as the tert-butylcarbamate using standard conditions. Subsequent N-debenzylation followed by reaction with 4-nitrophenyl chloroformate afforded carbamate 3 in good yield. Pyrrolidine 5 was prepared in two steps from commercially available (3R)-(-)-1-benzyl-3-aminopyrrolidine (4).7 Trifluoroacetamide protection followed by N-debenzylation afforded 5 in quantitative yield. Coupling of carbamate 3 and pyrrolidine 5 yielded the orthogonally protected urea core 6. Theoretically the two ends of the rAPU core could be functionalized in either order; however, intermediates first elaborated on the left-hand side were much easier to purify. Scheme 2 depicts the chosen synthetic route to compounds in Tables 1 and 2. Deprotection of the amine with trifluoroacetic acid, coupling to the appropriate carboxylic acid, followed by base promoted deprotection of the trifluoroacetamide afforded primary amines 7. N-alkylation via reductive amination afforded rAPUs 8. Alternatively, urea 6 deprotection followed by coupling to 5-bromo-2-thiophene carboxylic acid using EDCI/ HOBt afforded 9 in good yield. Trifluoroacetamide deprotection followed by reductive amination with 4,4dimethylcyclohexanone yielded the secondary amine 10. Suzuki coupling of bromothiophene 10 with a variety of arylboronic acids afforded rAPUs 11. The furan 12 was synthesized in an analogous manner starting with 5-bromo-2-furan-carboxylic acid.

For tertiary methyl amines, as exemplified by 14a–c, the methylation step was conveniently performed at an earlier point in the synthesis (Scheme 3). Trifluoroacetamide 6 was methylated with methyl iodide in quantitative yield. Carbamate deprotection, amide coupling, a second deprotection, and a final reductive amination afforded rAPUs 14. The trifluoroethyl derivative 17 was prepared in four steps from 6. Borane reduction of the trifluoroacetamide afforded amine 15. N-methylation using formaldehyde via reductive amination yielded tertiary amine 16. Deprotection followed by amide coupling afforded the desired trifluoroethyl derivative 17. Scheme 4 was employed when reductive amination was

not an option to prepare the tertiary amine of the right-hand side (23a-f). Alcohol 21 was prepared in four steps. Coupling of the arylcarboxylic acid to (3S)-(-)-1-benzyl-3-(methylamino)pyrrolidine (2) afforded 18. N-debenzylation of 18 yielded amine 19. Derivatization with 4-nitrophenyl chloroformate afforded carbamate 20. The urea core was synthesized via 4-nitrophenol displacement with (S)-(-)-3-hydroxypyrrolidine and afforded alcohol 21 in 32% (four steps). Reaction of 21 with methanesulfonyl chloride at low temperature afforded mesylate 22. Displacement of mesylate 22 with a variety of amines afforded rAPUs 23.8

The rAPUs described herein (Tables 1-3), obtained as single diastereomers, were tested in the MCH competitive binding assay. The functional antagonism was measured of all compounds with K_is less than 10 nM,⁵ in general a 3-fold reduction in potency was observed. Replacement of the biphenyl for phenyl thiophene in the rAPU series did not result in the same dramatic increase in potency as was achieved for the APU series (a 17-fold increase in binding affinity was observed for 1 compared to its biphenyl analog).4i Most of the rAPU phenyl thiophene analogs were up to 3-fold less active (Table 1). In addition, as demonstrated by 8a and 8b, compounds with para-ethyl or para-methoxy substitution were about 3-fold more potent than the trifluoromethyl analog 8c, suggesting that electron-donating substituents are favored.

Even though the introduction of the thiophene moiety did not generally result in a large increase in potency, it did result in highly potent compounds when combined with the gem dimethylcyclohexyl right-hand side described previously⁴¹ (as summarized in Table 2). Compounds **11a** and **11b**, with left-hand side *para*-ethyl and *para*-methoxy, respectively, highlight the importance of the *para* substitution. Both compounds have an affinity equal to 1 nM, which is significantly better relative to the unsubstituted analog, **11b**, with an affinity of 25 nM. Increasing the size of the *para* substituent also led to a drop in affinity with the ethoxy analog **11g** being 8-fold less potent than **11b**. This supports the hypothesis that the receptor binding pocket around the biaryl motif is sensitive to sterics. Two other promising analogs are

Scheme 1. Reagents and conditions: (a) di-tert-butyl dicarbonate, TEA, DCM, rt, 2 h, quantitative; (b) 10% Pd/C, ammonium formate, EtOH, Δ_R , 2 h, 76% (3) and quantitative (5); (c) 4-nitrophenyl chloroformate, TEA, THF, 0 °C to rt, 2 h, 68%; (d) ethyl trifluoroacetate, TEA, MeOH, rt, 12 h, quantitative; (e) TEA, DMF, 90 °C, 2.5 h, 66%.

Scheme 2. Reagents and conditions: (a) TFA, DCM, rt, 1 h, quantitative; (b) 5-bromo-2-thiophene carboxylic acid, EDCI, HOBt, TEA, DCM, rt, 18 h, 85%; (c) potassium carbonate, 10% aqueous EtOH, 80 °C, 16 h, 71–95%; (d) 4,4-dimethylcyclohexanone, Na(OAc)₃BH, MeOH, rt, 18 h, 69%; (e) arylboronic acid, Pd(PPh₃)₄, toluene, EtOH, 2 M sodium carbonate, 80 °C, 18 h, 38–50%; (f) biarylcarboxylic acid, EDCI, HOBt, TEA, DCM, rt, 16 h, 63–72%; (g) aldehyde or ketone, Na(OAc)₃BH or BH₃–pyridine, MeOH, rt, 12 h, 20–60%.

Table 1. Binding affinities of rAPUs 8a-f toward MCH-R1

Compound	Ar	K_i^a (nM) (p $K_i \pm SEM$)
8a		5.7 (8.3 ± 0.1)
8b		$4.9 (8.3 \pm 0.1)$
8c	F ₃ C	22 (7.7 ± 0.2)
8d	SX	$2.2 (8.7 \pm 0.1)$
8e	S X	21 (7.7 ± 0.1)
8f	F ₃ C	$60 \ (7.2 \pm 0.2)$

^a K_i values (n = 2-6).

the 2-methyl-4-methoxy (11e) and the benzodioxane (11d) compounds both of which had $K_i = 2 \text{ nM}$. Replacement of the thiophene with furan (11f and 12) led to a 1000-fold loss of potency. This loss in

potency could be attributed to one of two features of the furan ring: the increase in hydrophilicity or the decrease in ring size leading to a disfavored orientation of the pendant aryl ring.

Unfortunately, whilst the reversal of the aminopyrrolidine had enabled us to identify compounds with similar potency to the original APU series, it had a detrimental effect on CYP2D6 inhibition. rAPU 11f had a CYP2D6 IC50 of 900 nM compared to 14 μ M for compound 1. Indeed, with the exception of 11g, most of the compounds in this series showed low micromolar inhibition of CYP2D6 which remained a concern. It was anticipated that brain/plasma ratios would likely be below 1 for our initial compounds and thus higher doses would be required to achieve adequate brain levels for efficacy. The resulting peripheral concentration was expected to reach levels which would significantly inhibit CYP2D6.

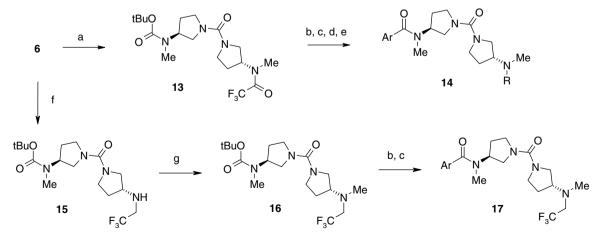
Utilizing the most potent left-hand side, the *p*-ethyl biaryls, a final round of optimization was designed. The issue of *CYP2D6* inhibition was addressed in an attempt to identify the most suitable compounds for further development (Table 3). Lipophilicity is well documented as being an important contributory factor in cytochrome P450 inhibition. Compounds **11a**—**h** have *c* log *P*s in the range of 5.0–6.8. Our first strategy was to investigate this property by reducing the size of the right-hand side substituent in order to reduce the *c* log *P*. An additional strategy was to investigate tertiary amines. One obvious difference between the APU series, which is devoid of *CYP2D6* inhibition issues, and the rAPU series is the nature of the terminal amine. In the

Table 2. Binding affinities of rAPUs 11a-h, 12 toward MCH-R1

Compound	Ar	$K_i^a (nM) (pK_i \pm SEM)$	$CYP2D6 \text{ IC}_{50}^{\text{ b}} (\mu\text{M})$ (pIC ₅₀ ± SEM)
11a	S	$0.7 (9.2 \pm 0.01)$	$3.2 (5.5 \pm 0.1)$
11b	o s x	$0.9 \ (9.0 \pm 0.1)$	$5.6 (5.3 \pm 0.1)$
11e	S X	$1.6~(8.8\pm0.04)$	$3.5 (5.5 \pm 0.1)$
11d	s	$1.9~(8.7\pm0.01)$	$4.5\ (5.4\pm0.1)$
11e	o s	$2.4~(8.6\pm0.2)$	$4.5\ (5.4\pm0.1)$
11f	F ₃ C S	$3.7~(8.4\pm0.00)$	$0.9 (6.1 \pm 0.1)$
11g	o s x	$16 \ (7.8 \pm 0.01)$	$12.5 (4.9 \pm 0.02)$
11h	S	25 (7.6 ± 0.2)	$2.3 (5.7 \pm 0.01)$
12	F ₃ C	$3600 (5.4 \pm 0.1)$	ND

^a K_i values (n = 2-6).

^b IC₅₀ values (n = 2 or 3). ¹⁴



Scheme 3. Reagents and conditions: (a) methyl iodide, potassium carbonate, DMF, 80 °C, 14 h, 98%; (b) TFA, DCM, rt, 1 h, quantitative; (c) biarylcarboxylic acid, EDCI, HOBt, TEA, DCM, rt, 16 h, 50–72%; (d) potassium carbonate, 10% aqueous MeOH, 70 °C, 16 h, quantitative; (e) aldehyde or ketone, Na(OAc)₃BH or BH₃–pyridine, MeOH, rt, 12 h, 20–60%; (f) BH₃, THF, Δ_R , 6 h, 94%, (g) HCHO_(aq), BH₃–pyridine, MeOH, rt, 16 h, 65%.

APU series this amine is exclusively tertiary by incorporation into the pyrrolidine ring, whereas in the rAPU series previous analogs had predominantly been second-

ary amines. It has been reported¹¹ that the presence of a hydrogen bond donor on the ligand is crucial for binding and inhibition potency of the *CYP*2D6 enzyme.

Scheme 4. Reagents and conditions: (a) carboxylic acid, HBTU, HOBt, DIEA, DMF, rt, 18 h; (b) 10% Pd/C, ammonium formate, EtOH, Δ_R , 2 h; (c) 4-nitrophenyl chloroformate, TEA, THF, 0 °C to rt, 2 h; (d) (S)-3-hydroxypyrrolidine hydrochloride, TEA, DMF, 80 °C, 4 h, 32% (four steps); (e) methanesulfonyl chloride, TEA, DCM, -40 °C, 1 h, 52-89%; (f) amine, DMA, 60-90 °C, 24 h, 30-50%.

Conversion to a tertiary amine, even though it would still be protonated at physiological pH, may alter the steric environment of the amine enough to prevent this key interaction.

The primary amine 7a had moderate affinity for the MCH-R1 receptor with a K_i of 18 nM. In addition, it did not inhibit CYP2D6. A significant increase in MCH-R1 affinity was observed with a more lipophilic secondary amine. Amines substituted with a saturated ring such as cyclopentyl (8g) or pyran (23a, 8h) gave rise to potent analogs, ≤5 nM. Branched acyclic side chains also had good binding affinity, for example 23e. Removing the cyclic moiety and replacing it with a less lipophilic substituent led to a drop in potency. For example, the methylamine analog 14c and the isopropyl analog 8i were 3- and 5-fold less potent than 8g, respectively. Extending the chain length by introducing a methylene linker led to a 4-fold drop in potency (compound 8k). As seen previously, some of the secondary amines exhibited significant CYP2D6 activity: the cyclopentyl compound 8g had a CYP2D6 IC₅₀ of 2.6 μM. An improvement to 5.1 and 5.8 μM was observed when a pyran, with or without branching, was introduced (23a and 8k, respectively), presumably due to the presence of the more hydrophilic oxygen atom. Small lipophilic substituents also decreased the CYP2D6 activity; the isopropyl (8i) and methyl (14c) secondary amines had IC₅₀s of 9.2 and 18.7 μM, respectively. Modification of the amines to give tertiary methylamines resulted in compounds with greatly improved CYP2D6 activity. For example, 23a and 14a both had K_{is} of around 2–3 nM and had CYP2D6 IC₅₀s equal to 5.1 and $>30 \mu M$, respectively. This is probably due to the removal of a favorable H-bonding interaction. Unfortunately, the tertiary amines were generally less metabolically stable than their parent secondary amines. We rationalized that reducing the basicity of the amine, either by the introduction of a β-oxygen atom or trifluoromethyl substituent, may lead to a drop in CYP2D6 activity. These effects have been observed previously;¹² for CYP2D6 substrates decreasing the pK_a of an amine from 9.5 to 4.5 by switching a β -meth-

yl to β-trifluoromethyl resulted in a 20-fold drop in the $CYP2D6 K_{\rm m}$. In our case we were pleased to observe the desired effect of decreasing the CYP2D6 activity, but the decreases in basicity also resulted in moderate to large drops in MCH-R1 binding affinity. The morpholine derivative 23d was a relatively poor MCH-R1 binder compared to its piperidine analog. The branched acyclic analog 23f retained some potency (14 nM) compared to 23e. However, incorporation of the hydroxyethyl moiety (compound 81) led to a further drop in potency to 32 nM. This could be attributed not only to the lower basicity of the amine, but also to increased hydrophilicity of the side chain. The most dramatic loss in potency was observed with the trifluoroethyl substituted analog 17 which had a binding affinity of 500 nM. This is probably due to the decreased basicity of the amine. The microsomal stability of compounds with ethyl and methoxy left-hand side substitution was lower than for para-trifluoromethyl. It was also apparent that there was a decrease in stability for phenyl thiophene containing compounds; identical substitutions had scaled intrinsic clearances of 83 mL/min/kg for phenyl thiophene 8i versus 22 mL/min/kg for its biphenyl analog 8j. The p-methoxy derivative 23b also had higher scaled intrinsic clearances (91 mL/min/kg) than 23a (55 mL/min/kg).

With the availability of a number of compounds having good potency, we addressed the brain penetration issue. One likely explanation for the low brain penetration in this series relative to the APU series is the presence of a secondary amine. Choosing compounds was complicated by the lower microsomal stability of the tertiary amines and thus brain/plasma levels were measured by iv cassette dosing. 13 Results for representative compounds tested are shown in Table 4. Secondary amines typically had low penetration as exemplified by 23a and 8h, and the addition of even a methyl group was enough to substantially improve the brain/plasma ratio, as shown with 14a. In the 14a cassette, we also measured the appearance of metabolite 23a in plasma and brain. The amount of 23a in plasma was significant, 2.4 ng/g compared to 6.7 ng/g for the parent compound, confirm-

Table 3. Binding affinities toward MCH-R1, CYP2D6 inhibition and predicted oral bioavailabilities of rAPUs

$$Ar \xrightarrow{N (S)} N \xrightarrow{N (R)} R^{R}$$

Compound	Ar	R ¹	R ²	$K_i^a (nM)$ (p $K_i \pm SEM$)	$CYP2D6 \text{ IC}_{50}^{\text{ b}}$ (μM) ($p\text{IC}_{50} \pm \text{SEM}$)	Scaled intrinsic clearance (mL/min/kg)
7a	J C	Н	Н	18 (7.7 ± 0.03)	>30	ND
8g	sty	\searrow	Н	$2 (8.8 \pm 0.01)$	$2.6 \ (5.6 \pm 0.1)$	60
8h		$\langle \rangle$	Н	5 (8.3 ± 0.1)	>10	18
8i	SX	\	Н	11 (8.0 ± 0.1)	$9.2\ (5.0\pm0.03)$	83
8j		\	Н	11 (8.0 ± 0.00)	$15.6 \ (4.8 \pm 0.03)$	22
8k	s	χ \circ	Н	$8 (8.1 \pm 0.1)$	$5.8 (5.2 \pm 0.03)$	43
81		У	Н	$32\ (7.5\pm0.1)$	>30	ND
14a	SX	$\langle \rangle$	Me	$3 (8.6 \pm 0.1)$	>30	210
14b	s	Me	Me	$5 (8.3 \pm 0.1)$	>30	240
14c	SX	Me	Н	$6 (8.2 \pm 0.1)$	$18.7 \ (4.7 \pm 0.04)$	55
17	SX	F F	Me	$500 \ (6.3 \pm 0.2)$	>30	ND
23a	sx		Н	$2 (8.6 \pm 0.1)$	$5.1 (5.3 \pm 0.02)$	55
23b	s x		Н	$10 \ (8.0 \pm 0.1)$	>30 (cont	91 inued on next page)

Table 3 (continued)

Compound	Ar	R^1	\mathbb{R}^2	$K_i^a \text{ (nM)}$ (p $K_i \pm \text{SEM}$)	$CYP2D6 IC_{50}^{b}$ (μ M) ($pIC_{50} \pm SEM$)	Scaled intrinsic clearance (mL/min/kg)
23c		, N	_	$4 (8.4 \pm 0.04)$	$10.4 \ (5.0 \pm 0.00)$	88
23d		, NH	_	$69\ (7.2\pm0.1)$	>30	ND
23e		\searrow	Н	$4 (8.4 \pm 0.1)$	$4.3~(5.4\pm0.01)$	150
23f		X-0-	Н	$14 \ (7.9 \pm 0.01)$	$24.5 \ (4.6 \pm 0.01)$	ND

ing that 14a was indeed metabolically unstable in rats. Brain levels were also measured and 23a was found to be 60× less abundant (at 0.35 ng/g) than 14a. The best compound, 23c, had comparable brain/plasma ratio and potency to 1 but the microsomal stability was marginal and the plasma and brain levels in rats were both about 4× lower than 1. We therefore placed this series on hold in favor of generating efficacy and toxicology data from the APU series.

In summary, we have thoroughly explored the scope of the rAPU series as potent and functional MCH-R1 antagonists. Extensive SAR was performed around the biarylcarboxamide, p-ethyl biphenyl and p-ethyl phenyl thiophene emerged as promising replacements for p-trifluoromethylbiphenyl. SAR carried out around the basic nitrogen moiety highlighted some interesting compounds. In particular, tertiary amines from this series exhibited desirable CYP2D6 activity and rat brain expo-

Table 4. B/P ratios of rAPUs iv cassette PK

Compound	Ar	R ¹	R ²	Average plasma concn ^a (ng/g)	Average brain concn ^a (ng/g)	B/P ratio ^a
8h			Н	14.5	5.1	0.4
14a	s		Me	6.7	21.8	3.4
23a	SX		Н	30.5	17.3	0.6
23c		$\langle N \rangle$	_	6.2	38.3	6.5

^a Average plasma and brain concentrations and brain/plasma ratios estimated by the cassette dosing procedure. ¹³

^a K_i values (n = 2-6). ⁹ ^b IC₅₀ values (n = 2 or 3). ¹⁴

sure and the best compound was the piperidine derivative 23c.

Acknowledgments

We are indebted to Mr. John Harman, Mr. Chris DeVore, and Mr. Shawn Ayube for LC–MS support. We also wish to thank Ms. Monica Mistry, Mr. Jason Haelewyn, and Mr. Rajesh Huntley for pharmacology support and Dr. John Saunders for valuable discussions. This work was partly supported by NIH Grant 2-R44-DK059107-02.

References and notes

- Vaughan, J. M.; Fischer, W. H.; Hoeger, C.; Rivier, J.; Vale, W. Endocrinology 1989, 125, 1660.
- (a) Boutin, J. A.; Suply, T.; Audinot, V.; Rodriguez, M.; Beauverger, P.; Nicolas, J.-P.; Galizzi, J.-P.; Fauchère, J.-L. Can. J. Physiol. Pharmacol. 2002, 80, 388, and references cited therein; (b) Sone, M.; Takahashi, K.; Murakami, O.; Totsune, K.; Arihara, Z.; Satoh, F.; Sasano, H.; Ito, H.; Mouri, T. Peptides 2000, 21, 245.
- (a) Takekawa, S.; Asami, A.; Ishihara, Y.; Terauchi, J.; Kato, K.; Shimomura, Y.; Mori, M.; Murakoshi, H.; Kato, K.; Suzuki, N.; Nishimura, O.; Fujino, M. Eur. J. Pharmacol. 2002, 438, 129; (b) Borowsky, B.; Durkin, M. M.; Ogozalek, K.; Marzabadi, M. R.; DeLeon, J.; Heurich, R.; Lichtblau, H.; Shaposhnik, Z.; Daniewska, I.; Blackburn, T. P.; Branchek, T. A.; Gerald, C.; Vaysse, P. J.; Forray, C. Nat. Med. 2002, 8, 825; (c) Souers, A. J.; Gao, J.; Brune, M.; Bush, E.; Wodka, D.; Vasudevan, A.; Judd, A. S.; Mulhern, M.; Brodjian, S.; Dayton, B.; Shapiro, R.; Hernandez, L. E.; Marsh, K. C.; Sham, H. L.; Collins, C. A.; Kym, P. R. J. Med. Chem. 2005, 48, 1318; (d) McBriar, M. D.; Guzik, H.; Xu, R.; Paruchova, J.; Li, S.; Palani, A.; Clader, J. W.; Greenlee, W. J.; Hawes, B. E.; Kowalski, T. J.; O'Neill, K.; Spar, B.; Weig, B. J. Med. Chem. 2005, 48, 2274; (e) Souers, A. J.; Gao, J.; Wodka, D.; Judd, A. S.; Mulhern, M. M.; Napier, J. J.; Brune, M. E.; Bush, E. N.; Brodjian, S. J.; Dayton, B. D.; Shapiro, R.; Hernandez, L. E.; Marsh, K. C.; Sham, H. L.; Collins, C. A.; Kym, P. R. Bioorg. Med. Chem. Lett. 2005, 15, 2752; (f) Huang, C. Q.; Baker, T.; Schwarz, D.; Fan, J.; Heise, C. E.; Zhang, M.; Goodfellow, V. S.; Markison, S.; Gogas, K. R.; Chen, T.; Wang, X.-C.; Saunders, J.; Zhu, Y.-F. Bioorg. Med. Chem. Lett. 2005, 15, 3701; (g) Vasudevan, A.; LaMarche, M. J.; Blackburn, C.; Che, J. L.; Luchaco-Cullis, C. A.; Lai, S.; Marsilje, T. H.; Patane, M. A.; Souers, A. J.; Wodka, D.; Geddes, B.; Chen, S.; Brodjian, S.; Falls, D. H.; Dayton, B. D.; Bush, E.; Brune, M.; Shapiro, R. D.; Marsh, K. C.; Hernandez, L. E.; Sham, H. L.; Collins, C. A.; Kym, P. R. Bioorg. Med. Chem. Lett. 2005, 15, 4174; (h) Palani, A.; Shapiro, S.; McBriar, M. D.; Clader, J. W.; Greenlee, W. J.; Spar, B.; Kowalski, T. J.; Farley, C.; Cook, J.; Van Heek, M.; Weig, B.; O'Neill, K.; Graziano, M.; Hawes, B. J. Med. Chem. 2005, 48, 4746; (i) Kym, P. R.; Iyengar, R.; Souers, A. J.; Lynch, J. K.; Judd, A. S.; Gao, J.; Freeman, J.; Mulhern, M.; Zhao, G.; Vasudevan, A.; Wodka, D.; Blackburn, C.; Brown, J.; Che, J. L.; Cullis, C.; Lai, S. J.; LaMarche, M. J.; Marsilie, T.; Roses, J.; Sells, T.; Geddes, B.; Govek, E.; Patane, M.; Fry, D.; Dayton, B. D.; Brodjian, S.; Falls, D.; Brune, M.; Bush, E.; Shapiro, R.; Knourek-Segel, V.; Fey, T.; McDowell, C.; Reinhart, G. A.; Preusser, L. C.;

- Marsh, K.; Hernandez, L.; Sham, H. L.; Collins, C. A. J. Med. Chem. 2005, 48, 5888; (j) Hervieu, G. Expert Opin. Ther. Targets 2003, 7, 495, and references cited therein.
- 4. (a) Carpenter, A. J.; Hertzog, D. L. Expert Opin. Ther. Patents 2002, 12, 1639; (b) Collins, C. A.; Kym, P. R. Curr. Opin. Invest. Drugs 2003, 4, 386; (c) Clark, D. E.; Higgs, C.; Wren, S. P.; Dyke, H. J.; Wong, M.; Norman, D.; Lockey, P. M.; Roach, A. G. J. Med. Chem. 2004, 47, 3962; (d) Arienzo, R.; Clark, D. E.; Cramp, S.; Daly, S.; Dyke, H. J.; Lockey, P.; Norman, D.; Roach, A. G.; Stuttle, K.; Tomlinson, M.; Wong, M.; Wren, S. P. *Bioorg*. Med. Chem. Lett. 2004, 14, 4099; (e) Souers, A. J.; Wodka, D.; Gao, J.; Lewis, J. C.; Vasudevan, A.; Gentles, R.; Brodjian, S.; Dayton, B.; Ogiela, C. A.; Fry, D.; Hernandez, L. E.; Marsh, K. C.; Collins, C. A.; Kym, P. R. Bioorg. Med. Chem. Lett. 2004, 14, 4873; (f) Vasudevan, A.; Wodka, D.; Verzal, M. K.; Souers, A. J.; Gao, J.; Brodjian, S.; Fry, D.; Dayton, B.; Marsh, K. C.; Hernandez, L. E.; Ogiela, C. A.; Collins, C. A.; Kym, P. R. Bioorg. Med. Chem. Lett. 2004, 14, 4879; (g) Souers, A. J.; Wodka, D.; Gao, J.; Lewis, J. C.; Vasudevan, A.; Gentles, R.; Brodjian, S.; Dayton, B.; Ogiela, C. A.; Fry, D.; Hernandez, L. E.; Marsh, K. C.; Collins, C. A.; Kym, P. R. Bioorg. Med. Chem. Lett. 2004, 14, 4883; (h) Receveur, J.-M.; Bjurling, E.; Ulven, T.; Little, P. B.; Nørregaard, P. K.; Högberg, T. Bioorg. Med. Chem. Lett. 2004, 14, 5075; (i) Grey, J.; Dyck, B.; Rowbottom, M. W.; Tamiya, J.; Vickers, T. D.; Zhang, M.; Zhao, L.; Heise, C. E.; Schwarz, D.; Saunders, J.; Goodfellow, V. S. Bioorg. Med. Chem. Lett. 2005, 15, 999; (j) Su, J.; McKittrick, B. A.; Tang, H.; Czarniecki, M.; Greenlee, W. J.; Hawes, B. E.; O'Neill, K. Bioorg. Med. Chem. 2005, 13, 1829; (k) Rowbottom, M. W.; Vickers, T. V.; Dyck, B.; Tamiya, J.; Zhang, M.; Zhao, L.; Grey, J.; Provencal, D.; Schwarz, D.; Heise, C. E.; Mistry, M.; Fisher, A.; Dong, T.; Hu, T.; Saunders, J.; Goodfellow, V. S. Bioorg. Med. Chem. Lett. 2005, 15, 3439; (1) Kanuma, K.; Omedera, K.; Nishiguchi, M.; Funakoshi, T.; Chaki, S.; Semple, G.; Tran, T.-A.; Kramer, B.; Hsu, D.; Casper, M.; Thomsen, B.; Beeley, N.; Sekiguchi, Y. Bioorg. Med. Chem. Lett. 2005, 15, 2565; (m) Kanuma, K.; Omedera, K.; Nishiguchi, M.; Funakoshi, T.; Chaki, S.; Semple, G.; Tran, T.-A.; Kramer, B.; Hsu, D.; Casper, M.; Thomsen, B.; Sekiguchi, Y. Bioorg. Med. Chem. Lett. 2005, 15, 3853; (n) Ulven, T.; Frimurer, T. M.; Receveur, J.-M.; Little, P. B.; Rist, Ø.; Nørregaard, P. K.; Högberg, T. J. Med. Chem. 2005, 48, 5684.
- Rowbottom, M. W.; Vickers, T. D.; Dyck, B.; Grey, J.; Tamiya, J.; Zhang, M.; Kiankarimi, M.; Wu, D.; Dwight, W.; Wade, W. S.; Saunders, J.; Schwarz, D.; Heise, C. E.; Madan, A.; Fisher, A.; Petroski, R.; Goodfellow, V. S. *Bioorg. Med. Chem. Lett.* Submitted for publication.
- 6. (3S)-(-)-1-Benzyl-3-(methylamino)pyrrolidine is commercially available from Tokyo Chemical Industry (TCI) America, Portland, Oregon. The e.e. was determined to be >97%.
- 7. (3*R*)-(-)-1-Benzyl-3-aminopyrrolidine is commercially available from Tokyo Chemical Industry (TCI) America, Portland, Oregon (e.e. 98%).
- 8. Typical experimental procedure for preparation of rAPUs via mesylate displacement: synthesis of **23c**. To a solution of the mesylate **22** (30 mg, 0.06 mmol) in DMA (0.5 mL) was added piperidine (100 μ L, 1 mmol). The reaction mixture was heated at 60 °C for 20 h then cooled to room temperature. The mixture was diluted to 1 mL with methanol and purified by preparative HPLC. Concentration of the pure fractions in vacuo afforded the TFA salt of **23c** (32 mg, 90%) as a colorless oil: ¹H NMR (CDCl₃ + one drop CD₃OD) δ 7.59 (d, 2H, J = 8.1 Hz),

- 7.48 (d, 2H, J = 8.4 Hz), 7.41 (d, 2H, J = 8.1 Hz), 7.25 (d, 2H, J = 8.4 Hz), 3.20–3.87 (m, 10H), 3.05 (m, 2H), 2.95 (m, 2H), 2.66 (q, 2H, J = 8.1 Hz), 1.63–2.42 (m, 10H), 1.23 (t, 3H, J = 8.1 Hz); MS (m/z) 489.3 (M+H) $^{+}$.
- 9. On each assay plate, a standard antagonist of comparable affinity to those being tested was included as a control for plate-to-plate variability. Overall K_i values were highly reproducible, the standard error of the mean (SEM) being reported. All compounds described were assayed in 2–6 independent experiments.
- Lewis, D. F. V.; Jacobs, M. N.; Dickins, M. D. *Drug Discovery Today* 2004, 12, 530.
- Hutzler, J. M.; Walker, G. S.; Wienkers, L. C. Chem. Res. Toxicol. 2003, 16, 450.

- Upthagrove, A. L.; Nelson, W. L. *Drug Metab. Dispos.* 2001, 29, 1377.
- 13. In the cassette dosing experiments a group of three male rats were dosed with four NCEs and a positive control. Plasma and brain samples were collected at appropriate intervals, processed, and analyzed by HPLC/MS. Calibration curves were generated and were used to quantify concentrations of the NCEs in plasma and brain. B/P ratios were determined and values >1 were presumed sufficient for the NCE to readily cross the BBB.
- 14. The *CYP*2D6 inhibition assay was carried out in the presence of the fluorescent substrate, AMMC. Quinidine was used as positive control. All compounds described with an $IC_{50} < 30 \mu M$ were assayed in 2 or 3 experiments.